Crystal Structure of 4-(4-Bromophenyl)-1,7,7-trimethyl-1,2,3,4,5,6,7,8-octahydroquinazoline-2,5-dione

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(Received December 25, 2000; Accepted April 16, 2001)

4-(4-Bromophenyl)-1,7,7-trimethyl-1,2,3,4,5,6,7,8-octahydroquinazoline-2,5-dione was obtained from Biginelli-type cyclocondensation¹ of 5,5-dimethylcyclohexane-1,3-dione with urea and 4-bromobenzaldehyde: yield 1.16 g (62.19%), m.p. 174 - 175°C. A summary of the key crystallographic information is given in Table 1. The atomic coordinates and equivalent isotropic displacement parameters with estimated standard deviations for atoms except H are listed in Table 2, and selected bond lengths and bond angles in Table 3. A perspective view of the title molecule showing the atomnumbering scheme is presented in Fig. 2.

The phenyl ring at C4 is nearly perpendicular to the

Table 1 Crystal and experimental data

Formula: $C_{17}H_{19}BrN_2O_2$ Formula weight = 363.25 Crystal system: monoclinic Space group: $P2_1/c$ $Z=4$ $a=11.233(1)$ Å $b=8.043(1)$ Å $\beta=95.62(1)$ ° c=17.594(2)Å $V=1581.9(3)$ Å ³ $D_c=1.525$ g/cm ³ μ (Cu K $_{\alpha}$) = 3.614 mm ⁻¹ T=295 K Yellow $F(0\ 0\ 0)=744$ Crystal size: $0.40\times0.40\times0.12$ mm $2\theta_{\max}=148.5$ ° R=0.051 Rw=0.136 No. of reflection used = 2745 $(I>2\sigma(I))$ No. of parameters = 203 Goodness-of-fit = 1.13 $(\Delta/\sigma)_{\max}=0.003$ $(\Delta\rho)_{\max}=0.076$ eÅ ⁻³ $(\Delta\rho)_{\min}=-0.04$ eÅ ⁻³
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$(\Delta \rho)_{\min} = -0.44 \text{ eÅ}^{-3}$
Measurements: Enraf-Nonius CAD-4 diffractometer
Refinement: full matrix least-squares (SHELXL-97)
Program system: CAD-4 EXPRESS software
Structure determination: SHELXS-97
Treatment of hydrogen atoms: geometric calculation

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quinazoline ring system, with the dihedral angle of 84.4(1)°. The observed bond lengths of both C-O in these structures are normal. The differences between the lengths of C2-N1 and C2-N3 are 0.065 Å, similar to the value of 0.065 Å found in 1,7,7-trimethyl-4-(4-methylphenyl)-1,2,3,4,5,6,7,8-octahydro-

Table 2 Final atomic coordinates and equivalent isotropic thermal parameters (U_{eq}) for non-hydrogen atoms

Atom	X	у	z	$U_{ m eq}/{ m \AA}^2$
Br1	0.1834(1)	0.8698(9)	0.0588(5)	0.0894(4)
O1	0.5042(3)	1.5551(5)	0.1276(2)	0.0583(9)
O2	0.8420(3)	1.0267(5)	0.2941(2)	0.0620(9)
N1	0.8420(3)	1.2179(5)	0.1981(2)	0.0479(8)
N3	0.6763(3)	1.1812(5)	0.2661(2)	0.0492(9)
C2	0.7878(4)	1.1329(5)	0.2555(3)	0.0461(10)
C4	0.5963(3)	1.2890(5)	0.2173(2)	0.0415(9)
C4a	0.6678(4)	1.3838(5)	0.1636(2)	0.0427(9)
C5	0.6062(4)	1.5108(6)	0.1170(3)	0.0455(9)
C6	0.6695(4)	1.5815(7)	0.0519(3)	0.0562(11)
C7	0.8035(4)	1.6041(6)	0.0733(3)	0.0477(10)
C8	0.8575(4)	1.4392(6)	0.1037(2)	0.0450(9)
C8a	0.7846(4)	1.3459(5)	0.1568(2)	0.0414(9)
C9	0.8655(5)	1.6531(8)	0.0023(3)	0.0654(14)
C10	0.8259(5)	1.7404(7)	0.1335(3)	0.0637(13)
C11	0.9629(4)	1.1645(7)	0.1846(3)	0.0588(12)
C12	0.4958(4)	1.1875(5)	0.1765(2)	0.0421(9)
C13	0.3791(4)	1.2153(5)	0.1908(3)	0.0464(9)
C14	0.2865(4)	1.1208(6)	0.1556(3)	0.0540(11)
C15	0.3117(4)	1.0005(6)	0.1054(3)	0.0540(11)
C16	0.4270(5)	0.9683(7)	0.0889(3)	0.0605(12)
C17	0.5181(4)	1.0633(6)	0.1238(3)	0.0531(11)

 $U_{\rm eq} = (1/3) \Sigma_{\rm i} \Sigma_{\rm j} U_{ij} a_i * a_j * (\boldsymbol{a}_i \cdot \boldsymbol{a}_j).$

Table 3 Selected geometric parameters (Å, °)

Br1 - C15	1.903 (5)	N3 – C2	1.342 (6)
O1 - C5	1.232 (5)	N3 – C4	1.465 (6)
O2 - C2	1.216 (6)	C4a – C8a	1.364 (6)
N1 - C2	1.407 (6)	C4 – C12	1.516 (6)
N1 - C8a	1.382 (6)	C7 – C9	1.540 (6)
N1 – C11	1.466 (6)	C7 – C10	1.527 (7)
C2-N1-C8a	121.8 (4)	C4a-C5-C6	117.4 (4)
C2-N3-C4	128.2 (3)	C5-C6-C7	112.4 (4)
N1-C2-N3	115.7 (3)	C6-C7-C8	109.3 (4)
N3-C4-C4a	109.5 (3)	N1-C8a-C8	116.4 (4)
C6-C7-C10	110.4 (4)	C8-C7-C9	108.0 (4)

Fig. 1 Chemical structure.

Table 4 Selected geometric parameters (Å, °)

D-H···A	D···A	D-H	Н…А	∠D-H···A		
C11-H11bO2 N3-H3O1 ⁱ	2.700 (7) 3.062 (5)	0.960 0.860	2.255 2.274	107.3 152.3		
Symmetry codes: (i) $-x+1,+y-1/2,-z+1/2$						

quinazoline-2,5-dione.² This situation can be attributed to the difference in the hybridization of the adjacent carbon atoms C8a and C4.

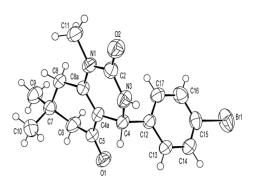


Fig. 2 The ORTEP drawing of the title compound with atom labeling. $\,$

An intermolecular hydrogen bond is found between N and O; geometric details are given in Table 4.

References

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- 2. M. M. Candan, E. Kendi, M. Yarim, S. Saraç, and M. Ertan, *Anal. Sci.*, **2000**, submitted.