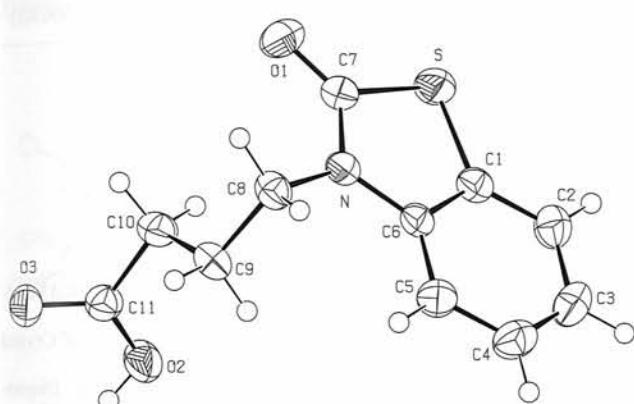


# Crystal structure of 4-(2-oxobenzothiazolin-3-yl)butanoic acid, C<sub>11</sub>H<sub>11</sub>NO<sub>3</sub>S

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**Abstract**

C<sub>11</sub>H<sub>11</sub>NO<sub>3</sub>S, triclinic,  $P\bar{1}$  (No. 2),  $a = 7.240(1)$  Å,  $b = 7.495(2)$  Å,  $c = 10.427(1)$  Å,  $\alpha = 83.89(1)^\circ$ ,  $\beta = 85.74(1)^\circ$ ,  $\gamma = 71.16(1)^\circ$ ,  $V = 532.0$  Å<sup>3</sup>,  $Z = 2$ ,  $R_{\text{gt}}(F) = 0.035$ ,  $wR_{\text{ref}}(F^2) = 0.096$ ,  $T = 293$  K.

**Source of material**

For the synthesis, 10.0 mmol ethyl 4-(2-oxobenzothiazolin-3-yl)butanoate in concentrated hydrochloric acid (50 ml) was stirred at room temperature for 2 hours, then refluxed for 4 hours. The reaction mixture was cooled, poured into 100 g ice-water, and stirred for 1 hour. The precipitate was collected by filtration, washed with water, dried and crystallised from water.

**Discussion**

The benzene ring defined by C1–C2–C3–C4–C5–C6 atoms is planar. Torsion angles of C1–C2–C3–C4, C6–C5–C4–C3 and C1–C6–C5–C4 are 0.4(3)°, 0.1(3)° and 0.3(3)°, respectively. The thiazolin ring defined by C1–C6–N–C7–S atoms is planar. Torsion angles of C7–S–C1–C6 and C6–N–C7–S are 0.6(1)° and 1.2(2)°, respectively. The dihedral angle between these two planes is 0.99(8)° implying that they are co-planar. The torsion angles of C7–N–C6–C5, N–C6–C1–C2, C7–S–C1–C2 and S2–C1–C2–C3 are 178.8(2)°, 179.2(2)°, -178.6(2)° and 179.1(2)°, respectively. O1 and C8 atoms are also at this plane. Torsion angles of C6–N–C7–O1, C1–S–C7–O1, C8–N–C6–C1 and C8–N–C7–O1 are -179.0(2)°, 179.2(2)°, -179.0(2)° and -0.7(3)°, respectively. C9 atom lies below 1.351(3) Å from thiazolin plane. Torsion angles of C7–N–C8–C9, C6–N–C8–C9 and N–C8–C9–C10 are 103.8(2)°, -78.0(2)° and -68.5(2)°, re-

spectively. The bond lengths and angles in the 4-(2-oxobenzothiazolin-3-yl)butanoic acid are all in accord with similar structures in the literature [1–3]. The bond lengths of C–C are between 1.378(3) Å – 1.522(3) Å. The bond lengths of C7=O1, C11–O2 and C11=O3 are 1.214(2) Å, 1.262(2) Å and 1.263(2) Å, respectively. The bond lengths of C7–S and C1–S are 1.782(2) Å and 1.743(2) Å, respectively. The bond lengths of C6–N, C7–N and C8–N are 1.395(2) Å, 1.368(2) Å and 1.464(2) Å, respectively.

**Table 1.** Data collection and handling.

Crystal:	white needle, size 0.06 × 0.27 × 0.48 mm
Wavelength:	Mo $K_\alpha$ radiation (0.71073 Å)
$\mu$ :	2.94 cm <sup>-1</sup>
Diffractometer, scan mode:	Enraf-Nonius CAD-4, $\omega/2\theta$
$2\theta_{\text{max}}$ :	52.58°
$N(hkl)$ measured, $N(hkl)$ unique:	2300, 2130
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$ , 1786
$N(\text{param})_{\text{refined}}$ :	147
Programs:	SHELXS-97 [4] SHELXL-97 [5], ORTEP-III [6]

**Table 2.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	x	y	z	$U_{\text{iso}}$
H(2)	2i	-0.4244	-0.0414	0.5929	0.070
H(8A)	2i	-0.1139	0.5419	0.6207	0.041
H(8B)	2i	-0.2521	0.5945	0.7436	0.041
H(5)	2i	-0.3345	0.4424	0.9429	0.046
H(4)	2i	-0.3352	0.3370	1.1597	0.053
H(9A)	2i	-0.4118	0.4711	0.6134	0.042
H(9B)	2i	-0.3442	0.3186	0.7301	0.042
H(10A)	2i	-0.1531	0.2997	0.4823	0.048
H(10B)	2i	-0.0717	0.1516	0.5994	0.048
H(2A)	2i	0.2404	0.0690	1.1354	0.048
H(3)	2i	-0.0519	0.1549	1.2552	0.054

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**Table 3.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>11</sub>	<i>U</i> <sub>22</sub>	<i>U</i> <sub>33</sub>	<i>U</i> <sub>12</sub>	<i>U</i> <sub>13</sub>	<i>U</i> <sub>23</sub>
S	2 <i>i</i>	0.32516(6)	0.16590(7)	0.85628(5)	0.0272(2)	0.0507(3)	0.0421(3)	-0.0099(2)	-0.0009(2)	-0.0008(2)
O(2)	2 <i>i</i>	-0.3456(2)	0.0043(2)	0.6171(1)	0.0589(9)	0.0509(9)	0.0424(8)	-0.0339(7)	-0.0150(6)	0.0080(6)
O(3)	2 <i>i</i>	-0.3910(2)	0.1564(2)	0.4210(1)	0.0590(9)	0.0510(8)	0.0337(7)	-0.0311(7)	-0.0082(6)	0.0022(6)
O(1)	2 <i>i</i>	0.2322(2)	0.3345(2)	0.6211(1)	0.0448(8)	0.069(1)	0.0399(8)	-0.0219(7)	0.0058(6)	0.0044(7)
N	2 <i>i</i>	-0.0148(2)	0.3698(2)	0.7785(1)	0.0288(7)	0.0354(8)	0.0317(8)	-0.0138(6)	-0.0040(6)	-0.0008(6)
C(6)	2 <i>i</i>	-0.0443(2)	0.3184(2)	0.9091(2)	0.0309(8)	0.0297(8)	0.0310(8)	-0.0143(7)	-0.0032(7)	-0.0042(7)
C(1)	2 <i>i</i>	0.1269(2)	0.2062(2)	0.9673(2)	0.0297(8)	0.0310(9)	0.0354(9)	-0.0108(7)	-0.0014(7)	-0.0053(7)
C(11)	2 <i>i</i>	-0.3165(3)	0.1242(3)	0.5303(2)	0.0350(9)	0.0347(9)	0.0354(9)	-0.0138(7)	0.0006(7)	-0.0047(7)
C(8)	2 <i>i</i>	-0.1711(3)	0.4898(2)	0.6966(2)	0.0362(9)	0.0313(9)	0.0382(9)	-0.0133(7)	-0.0091(7)	0.0015(7)
C(5)	2 <i>i</i>	-0.2193(3)	0.3681(3)	0.9809(2)	0.0314(9)	0.043(1)	0.039(1)	-0.0094(8)	0.0010(7)	-0.0052(8)
C(4)	2 <i>i</i>	-0.2187(3)	0.3048(3)	1.1104(2)	0.043(1)	0.049(1)	0.042(1)	-0.0148(9)	0.0107(8)	-0.0088(9)
C(9)	2 <i>i</i>	-0.2984(2)	0.3820(3)	0.6545(2)	0.0300(9)	0.036(1)	0.041(1)	-0.0119(7)	-0.0069(7)	-0.0038(8)
C(10)	2 <i>i</i>	-0.1901(3)	0.2362(3)	0.5611(2)	0.039(1)	0.048(1)	0.042(1)	-0.0243(9)	0.0025(8)	-0.0088(9)
C(7)	2 <i>i</i>	0.1740(3)	0.3046(3)	0.7310(2)	0.0339(9)	0.040(1)	0.0361(9)	-0.0184(8)	-0.0007(7)	-0.0032(7)
C(2)	2 <i>i</i>	0.1258(3)	0.1438(3)	1.0969(2)	0.042(1)	0.040(1)	0.037(1)	-0.0109(8)	-0.0083(8)	0.0006(8)
C(3)	2 <i>i</i>	-0.0487(3)	0.1949(3)	1.1678(2)	0.058(1)	0.047(1)	0.0299(9)	-0.018(1)	0.0018(8)	-0.0003(8)

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