# Synthesis and Crystal Structure of Hexahydrobis[(1,3-p-dimethylamino-benzyl)-1,3-diazepine]-2-selenone, $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{4} \mathrm{Se}$ 

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In the title compound, $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{4} \mathrm{Se}$, was synthesized and its crystal structure was determined by single crystal X-ray diffraction. The compound crystallizes in orthorhombic system, space group $P b c a, a=10.9960(10) \AA, b=14.9460(9) \AA$, $c=27.565(5) \AA$ and $Z=2$. According to X-ray crystallographic studies, the diazepine ring is observed to be in a disordered state. The site-occupancy of the major compenent refined to $0.53(1)$. The $\mathrm{C}=\mathrm{Se}$ bond length of $1.862(4) \AA$ is a double bond character. The dihedral angle between the two phenyl rings is $37.7(2)^{\circ}$.
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The use of exobycyclic alkenes of the type $\mathbf{1}(m=2$ or 3 ) as sources of $N$-heterocyclic carbene complexes of transition metals is well-established. ${ }^{1}$ However, larger rings, i.e. $m \geq 4$, have so far not been tackled. In an effort to fill the gap, we prepared 1c $\left(\mathrm{R}=p-\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NMe}_{2}\right.$; the R group was chosen so as to contain a quaternizable $-\mathrm{NMe}_{2}$ functionality which would increase the water solubility). Under mild conditions, 1c reacted with Group VI elements to give cyclic chalcogeno ureas (Eq. (1)). This behavior is typical of reactive tetraaminoalkenes $\mathbf{1 a}$ and 1b. ${ }^{2,3}$ In sharp contrast, $\mathbf{1 c}$ is inert to electrophilic transition metal complexes such as $\left[\mathrm{RuCl}_{2}\left(\eta^{6} \text {-arene }\right)\right]_{2}$. This remarkable difference aroused our curiosity. Therefore, the Xray structure of the seleno urea $\mathbf{2 c}$, derived from the alkene $\mathbf{1 c}$ was determined.
In a typical reaction, bis[1,3-bis( $p$-dimethylaminobenzyl)-hexahydro-1,3-diazepine-2-ylidene] (1c: $0.886 \mathrm{~g}, 1.00 \mathrm{mmol}$ ) was heated with elemental selenium $(0.200 \mathrm{~g}, 2.53 \mathrm{mmol})$ in refluxing toluene $(20 \mathrm{~mL})$ for 2 h . The resulting solution was cooled to room temperature and then filtered to remove the excess selenium. The volume of the filtrate was reduced to $c a$. 10 mL and hexane ( 10 mL ) was added. Upon cooling the solution to $-20^{\circ} \mathrm{C}$, we obtained cream crystals of the title compound $\left(0.54 \mathrm{~g}, 61 \%\right.$, m.p. $\left.=141-142^{\circ}\right)$. All reactions were carried out under argon atmosphere with the use of Schlenk techniques. The solvents were dried and deoxygenated by standard procedures. The new compound exhibited satisfactory spectroscopic and analytical data. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 1.34$ (br, $4 \mathrm{H}, \mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}$ ), 2.92 ( s, $\left.12 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right), 3.28$ (br, $4 \mathrm{H}, \mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}$ ), 5.07 (s, $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NMe}_{2}$ ), 6.67 (d, $\left.4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NMe}_{2}, J=8.6 \mathrm{~Hz}\right), 7.28\left(\mathrm{~d}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NMe}_{2}, J\right.$ $=8.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 24.30\left(\mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right.$ $\left.\mathrm{CH}_{2} \mathrm{~N}\right), 40.50\left(\mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \quad \mathrm{CH}_{2} \mathrm{~N}\right)$, $50.41\left(\mathrm{C}_{6} \mathrm{H}_{4} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right)$, $61.5 \quad\left(\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NMe}_{2}\right), \quad 112.4, \quad 125.0, \quad 129.6, \quad 150.0$ $\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{4} \mathrm{NMe}_{2}, 192.8(\mathrm{C}=\mathrm{Se})$. Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{4} \mathrm{Se}: \mathrm{C}$, $62.29 ; \mathrm{H}, 7.27 ; \mathrm{N}, 12.63$; found: C, $60.15 ; \mathrm{H}, 7.30 ; \mathrm{N}, 12.39$. The crystal and molecular structures of $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{4} \mathrm{Se}$ were

[^0]determined by single crystal X-ray diffraction. The crystal data, details of the data collection and the structure determination are given in Table 1. The chemical reaction of the title compound is shown in Fig. 1. The structure was solved by direct methods and refined by full-matrix least squares. The non-hydrogen atoms were refined anisotropically. The diazepine ring proved to be disordered over two orientations. The site-occupancy of the major disordered component refined to $0.53(1)$. The relatively high R values are related to the weak scattering that is most probably a consequence of the disorder in the diazepine ring. H atoms were added at calculated positions and refined using a riding model with $U_{\text {iso }}(\mathrm{H})=x U_{\text {eq }}($ parent atom), where $x$

Table 1 Crystal and experimental data

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Fig. 1 The chemical reaction of the title compound.

Table 2 Final coordinates and equivalent isotropic displacement parameters of the non- H atoms for the title compound

| Atom | $x$ | $y$ | $z$ | $U_{\text {eq }}\left(\AA^{2}\right)$ |
| :--- | :--- | :--- | :--- | :--- |
|  | $y$ |  |  |  |
| Se1 | $0.26058(4)$ | $0.36493(4)$ | $0.32121(2)$ | $0.0919(3)$ |
| N1 | $0.3764(4)$ | $0.2036(3)$ | $0.2977(1)$ | $0.111(2)$ |
| N2 | $0.5954(5)$ | $0.1797(4)$ | $0.5170(2)$ | $0.127(2)$ |
| N3 | $0.3748(4)$ | $0.3069(3)$ | $0.2380(1)$ | $0.097(2)$ |
| N4 | $0.0726(5)$ | $0.5245(3)$ | $0.0728(2)$ | $0.120(2)$ |
| C1 | $0.3448(4)$ | $0.2849(3)$ | $0.2818(2)$ | $0.085(2)$ |
| *C2A | $0.414(2)$ | $0.2597(10)$ | $0.1958(5)$ | $0.108(6)$ |
| \#C2B | $0.368(1)$ | $0.2252(12)$ | $0.1985(5)$ | $0.070(4)$ |
| *C3A | $0.4906(10)$ | $0.1889(6)$ | $0.1932(3)$ | $0.086(3)$ |
| \#C3B | $0.379(2)$ | $0.1651(10)$ | $0.2029(5)$ | $0.102(6)$ |
| \#C4A | $0.538(1)$ | $0.168(1)$ | $0.2439(7)$ | $0.118(7)$ |
| *C4B | $0.467(2)$ | $0.1114(8)$ | $0.2345(5)$ | $0.103(5)$ |
| *C5A | $0.5074(11)$ | $0.1728(8)$ | $0.2760(5)$ | $0.081(4)$ |
| \#C5B | $0.444(2)$ | $0.1295(10)$ | $0.2756(6)$ | $0.100(6)$ |
| C6 | $0.3221(6)$ | $0.1624(4)$ | $0.3404(2)$ | $0.105(2)$ |
| C7 | $0.4015(4)$ | $0.1659(3)$ | $0.3852(2)$ | $0.082(2)$ |
| C8 | $0.4880(5)$ | $0.2298(3)$ | $0.3929(2)$ | $0.097(2)$ |
| C9 | $0.5538(5)$ | $0.2348(3)$ | $0.4356(2)$ | $0.099(2)$ |
| C10 | $0.5327(5)$ | $0.1765(3)$ | $0.4732(2)$ | $0.098(2)$ |
| C11 | $0.4446(5)$ | $0.1109(3)$ | $0.4657(2)$ | $0.091(2)$ |
| C12 | $0.3827(5)$ | $0.1055(3)$ | $0.4219(2)$ | $0.090(2)$ |
| C13 | $0.6864(7)$ | $0.2486(5)$ | $0.5235(3)$ | $0.158(4)$ |
| C14 | $0.5458(10)$ | $0.1422(5)$ | $0.5601(3)$ | $0.177(5)$ |
| C15 | $0.3948(4)$ | $0.3985(4)$ | $0.2211(2)$ | $0.096(2)$ |
| C16 | $0.3088(4)$ | $0.4323(3)$ | $0.1829(2)$ | $0.078(2)$ |
| C17 | $0.3518(4)$ | $0.4797(3)$ | $0.1430(2)$ | $0.091(2)$ |
| C18 | $0.2753(4)$ | $0.5119(3)$ | $0.1074(2)$ | $0.095(2)$ |
| C19 | $0.1495(4)$ | $0.4959(3)$ | $0.1097(2)$ | $0.086(2)$ |
| C20 | $0.1069(4)$ | $0.4484(3)$ | $0.1496(2)$ | $0.083(2)$ |
| C21 | $0.1847(4)$ | $0.4181(3)$ | $0.1852(2)$ | $0.080(2)$ |
| C22 | $-0.0556(6)$ | $0.5136(5)$ | $0.0789(2)$ | $0.131(3)$ |
| C23 | $0.1135(7)$ | $0.5948(5)$ | $0.0417(3)$ | $0.155(3)$ |

$U_{\text {eq }}=(1 / 3) \sum_{i} \Sigma_{j} U_{i j}\left(a_{i}^{*} a_{j}^{*}\right)\left(\boldsymbol{a}_{i} \cdot \boldsymbol{a}_{j}\right)$.
*Site occupancy factor 0.53 . "Site occupancy factor 0.47.
$=1.5$ for methyl and 1.2 for others. The refined atomic parameters with equivalent isotropic temperature factors for non-hydrogen atoms are given in Table 2. Selected bond distances and bond angles are given in Table 3 and an ORTEP III drawing of the title compound together with the atomic numbering is shown in Fig. 2.
A notable feature of the ${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{2 c}$ is a broad signal at $\delta 192.8 \mathrm{ppm}$ assigned to $\mathrm{C}=\mathrm{Se}$ group and shifted considerably to high frequency compared to previously known cyclic seleno ureas. This is in agreement with the substantial $\mathrm{C}=\mathrm{Se}$ double bond $\left[1.862(4) \AA\right.$ A.$^{4}$ The displacement parameters of atoms $\mathrm{C} 2, \mathrm{C} 3, \mathrm{C} 4$ and C 5 of the diazepine ring are quite high, resulting in unusual bond lengths involving these atoms. As a result of the disorder, the average values of the N3-C2, C2-C3, $\mathrm{C} 3-\mathrm{C} 4, \mathrm{C} 4-\mathrm{C} 5$ and C5-N1 are 1.49(2), 1.13(2), 1.53(2), 1.07(2) and $1.56(1) \AA$, respectively. The distances between these partially populated sites of the C2A-C2B, C3A-C3B, C4A-C4B and C5A-C5B are $0.73(3), 1.30(2), 1.17(2)$ and $0.95(2) \AA$, respectively. The other bond lengths and angles are in agreement with those in the literature. The two partially populated sites of the diazepine ring are nonplanar.
The dihedral angle between the two phenyl rings is $37.7(2)^{\circ}$ and each of the dimethylamino groups is tilted from the phenyl

Table 3 Selected bond lengths ( $\AA$ ) and bond angles $\left({ }^{\circ}\right)$ for $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{4} \mathrm{Se}$

|  |  |  |  |
| :--- | :--- | :--- | :--- |
| Se1-C1 | $1.862(4)$ | $\mathrm{N} 3-\mathrm{C} 2 \mathrm{~A}$ | $1.38(2)$ |
| $\mathrm{N} 1-\mathrm{C} 1$ | $1.337(6)$ | $\mathrm{N} 3-\mathrm{C} 15$ | $1.444(7)$ |
| $\mathrm{N} 1-\mathrm{C} 5 \mathrm{~A}$ | $1.63(1)$ | $\mathrm{N} 3-\mathrm{C} 2 \mathrm{~B}$ | $1.60(2)$ |
| $\mathrm{N} 1-\mathrm{C} 6$ | $1.456(7)$ | $\mathrm{N} 4-\mathrm{C} 19$ | $1.388(7)$ |
| $\mathrm{N} 1-\mathrm{C} 5 \mathrm{~B}$ | $1.46(7)$ | $\mathrm{N} 4-\mathrm{C} 23$ | $1.429(9)$ |
| $\mathrm{N} 2-\mathrm{C} 10$ | $1.391(8)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.515(7)$ |
| $\mathrm{N} 3-\mathrm{C} 1$ | $1.351(6)$ | $\mathrm{C} 15-\mathrm{C} 16$ | $1.503(7)$ |
|  |  |  |  |
| C1-N1-C5A | $111.6(5)$ | $\mathrm{C} 1-\mathrm{N} 3-\mathrm{C} 2 \mathrm{~B}$ | $114.0(7)$ |
| $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $123.5(4)$ | $\mathrm{C} 15-\mathrm{C} 16-\mathrm{C} 17$ | $120.8(4)$ |
| $\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 5 \mathrm{~B}$ | $133.1(7)$ | $\mathrm{C} 2 \mathrm{~A}-\mathrm{N} 3-\mathrm{C} 15$ | $102.4(7)$ |
| $\mathrm{C} 5 \mathrm{~A}-\mathrm{N} 1-\mathrm{C} 6$ | $122.7(6)$ | $\mathrm{C} 2 \mathrm{~B}-\mathrm{N} 3-\mathrm{C} 15$ | $123.5(6)$ |
| $\mathrm{N} 2-\mathrm{C} 10-\mathrm{C} 9$ | $123.3(5)$ | $\mathrm{C} 22-\mathrm{N} 4-\mathrm{C} 23$ | $117.7(5)$ |
| $\mathrm{C} 5 \mathrm{~B}-\mathrm{N} 1-\mathrm{C} 6$ | $102.9(7)$ | $\mathrm{Sel}-\mathrm{C} 1-\mathrm{N} 3$ | $120.8(3)$ |
| $\mathrm{C} 13-\mathrm{N} 2-\mathrm{C} 14$ | $116.2(6)$ | $\mathrm{N} 1-\mathrm{Cl}-\mathrm{N} 3$ | $117.7(4)$ |
| $\mathrm{C} 1-\mathrm{N} 3-\mathrm{C} 2 \mathrm{~A}$ | $134.7(7)$ | $\mathrm{N} 3-\mathrm{C} 2 \mathrm{~A}-\mathrm{C} 3 \mathrm{~A}$ | $129.6(12)$ |
| $\mathrm{N} 3-\mathrm{C} 15-\mathrm{C} 16$ | $114.8(4)$ | $\mathrm{N} 1-\mathrm{C} 5 \mathrm{~A}-\mathrm{C} 4 \mathrm{~B}$ | $100.9(9)$ |



Fig. 2 An ORTEP drawing of the title compound showing the atomic numbering scheme. Displacement ellipsoids of non-H atoms are shown at the $30 \%$ probability level; H atoms are shown as small spheres of arbitrary size. H atoms of the diazepine ring have been omitted for clarity.
ring to which it is attached. The first dimethylamino group (C13-N2-C14) is tilted 22.3(6) ${ }^{\circ}$ from the plane of its phenyl ring, while the second dimethylamino group (C22-N4-C23) is tilted $21.9(7)^{\circ}$ from the plane of its phenyl ring. There are three intramolecular interactions: two of them are C-H...Se type and the other one is C-H $\cdots \mathrm{N}$ type [C6 $\cdots \mathrm{Se} 1$ 3.146(6), C6-H6B $0.970(7) \AA$ and $\mathrm{C} 6-\mathrm{H} 6 \mathrm{~B} \cdots \mathrm{Se} 1 \mathrm{Cl} 109.6(3)^{\circ}$; C15 $\cdots$ Sel 3.168(5), C15-H15B 0.970(5) $\AA$ and C15-H15B $\cdots$ Se1 113.9(3) ${ }^{\circ}$; C8…N1 $2.924(6), \mathrm{C} 8-\mathrm{H} 8 \quad 0.930(5) \AA$ and $\mathrm{C} 8-\mathrm{H} 8 \cdots \mathrm{~N} 11^{2} 00.6(3)^{\circ} \mathrm{J}$. The molecules in the crystal are packed at normal van der Waals distances.

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[^1]:    Formula: $\mathrm{C}_{23} \mathrm{H}_{32} \mathrm{~N}_{4} \mathrm{Se}$
    Formula weight $=443.49$
    Crystal System: orthorhombic
    Space group: Pbca $\quad Z=8$
    $T=293(2) \mathrm{K}$
    $a=10.9960(10) \AA$
    $b=14.9460(9) \AA$
    $c=27.565(5) \AA$
    $V=4530.2(9) \AA^{3}$
    $D_{\mathrm{x}}=1.301 \mathrm{~g} \mathrm{~cm}^{-3}$
    $\mu\left(\mathrm{Mo} \mathrm{K}_{\alpha}\right)=1.673 \mathrm{~mm}^{-1}$
    Crystal size: $(0.09 \times 0.40 \times 0.40) \mathrm{mm}^{3}$
    Radiation $\left(\mathrm{Mo} \mathrm{K}_{\alpha}\right)=0.71069 \AA$
    Reflection collected $=4617$
    Independent reflections $[I>2 \sigma(I)]=2626$
    No. of parameters: 291
    $R=0.061 ; w R=0.18$
    $w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.1185 P)^{2}+0.1331 P\right]$ where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right)$
    $(\Delta / \sigma)_{\max }=0.005$
    $(\Delta \rho)_{\text {max }}=0.49 \mathrm{e}^{-3}$
    $(\Delta \rho)_{\text {min }}=-0.55 \mathrm{e}^{-3}$
    Measurement: Enraf Nonius CAD-4 Diffractometer
    Program system: CAD-4 EXPRESS Software
    Structure determination: Direct method with SHELXS-97
    Refinement: Full matrix least-squares with SHELXL-97 in WINGX ${ }^{5}$

